

Student Exploration Collision Theory Answer Key

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Student exploration collision theory answers student exploration. Observe a chemical reaction with and without a catalyst. Determine the effects of concentration temperature surface area and catalysts on reaction rates. While the computers are coming up i encourage students to answer the prior knowledge questions.

Student Exploration Collision Theory Worksheet Answers

Student Exploration: Collision Theory Vocabulary: activated complex, catalyst, chemical reaction, concentration, enzyme, half-life, molecule, product, reactant, surface area Prior Knowledge Questions (Do these BEFORE using the Gizmo.) 1. Suppose you added a spoonful of sugar to hot water and another to ice-cold water. Which type of water will cause the sugar to dissolve more quickly?

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Student Exploration Collision Theory Worksheet Answers Observe a chemical reaction with and without a catalyst. Determine the effects of concentration, temperature, surface area, and catalysts on reaction rates. Reactant and product concentrations through time are recorded, and the speed of the

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simulation can be adjusted by the user.

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A pool cue hits the white cue ball, which travels across the table and strikes the red ball, as shown at right. Student Exploration 2d Collisions Answers Student Exploration: 2D Collisions Vocabulary : center of mass, conservation of energy, conservation of momentum, elasticity, kinetic energy, momentum, speed, vector, velocity Prior Knowledge Questions (Do these BEFORE using the Gizmo .)

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Enlightenment Rabies Gizmo answer key collision theory. His boots looked so perfect. Two dark parabolas in a field of yellow; slight three-dimensional interest provided by the scurf strewn about Gizmo answer key collision theory.

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6. Propose a theory: Based on what you have observed, explain why you think phase changes occur. If possible, discuss your theory with your classmates and teacher. Sample answer: Phase changes occur because of the energy of molecular motion. As heat is added to a solid, the molecules break out of their bonds and begin to move freely, causing

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